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ANSWER KEY 1 Modern chemistry answer key chapter 2. a 2. b 3. c 4. c 5. a 6. b 7 Modern chemistry answer key chapter 2. c 8.

Polarity in a water molecule is caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or basic.

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scientific method. hypothesis.  
model. theory. a logical approach  
to solving problems by observing  
and collec.... purposed answer  
than can be tested. an  
explanation of how phenomena  
occurred and how data or  
eleme.... is a broad generalization  
that explains a body of facts or  
phe....

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CHAPTER 2 REVIEW

Measurements and Calculations

SECTION 2 SHORT ANSWER

Answer the following questions in  
the space provided. 1. Complete  
the following conversions: a. 100  
mL 0.1 L b. 0.25 g 25 cg c. 400

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Answers  
cm<sup>3</sup> 0.4 L d. 400 cm<sup>3</sup> 0.0004 m

2. For each measuring device shown below, identify the quantity measured and tell when it would remain constant ...

## 2 Measurements and Calculations

### CHAPTER 2 REVIEW

#### Measurements and Calculations

#### SECTION 2-3 SHORT ANSWER

Answer the following questions in the space provided. 1. Report the number of significant figures in each of the following values: a. 0.002 37 g d. 64 mL b. 0.002 037 g e. 1.3 102 cm c. 350. J f. 1.30 102 cm 2. Write the value of the following operations using scientific notation: a. 103 1 0 1 2 0 6 b. 8 2 1 1 0



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## 2 Measurements and Calculations

$2\text{(g)} \rightarrow \text{NO}_2\text{(g)}$  a. Balance the equation.  $2\text{N}_2\text{O(g)} + 3\text{O}_2\text{(g)} \rightarrow 4\text{NO}_2\text{(g)}$  4 mol  $\text{NO}_2$ :3 mol  $\text{O}_2$  b. What is the mole ratio of  $\text{NO}_2$  to  $\text{O}_2$ ? 15.0 mol c. If 20.0 mol of  $\text{NO}_2$  form, how many moles of  $\text{O}_2$  must have been consumed? True d. Twice as many moles of  $\text{NO}_2$  form as moles of  $\text{N}_2\text{O}$  are consumed. True or False? False e. Twice as many grams of  $\text{NO}_2$  ...

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PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 8. 9.7 1014 Hz What is the frequency of light that has a wavelength of 310 nm? 9. 9.4 910 m What is the wavelength of electromagnetic radiation if its frequency is 3.2 10<sup>2</sup> Hz? Wavelength Wavelength Wave A Wave B 26

ARRANGEMENT OF ELECTRONS IN ATOMS ...

## **4 Arrangement of Electrons in**

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## **Atoms**

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided.

1. a The notation for sodium

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chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) ...

## 6 Chemical Bonding

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER

Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in

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## **5 The Periodic Law**

Modern Chemistry CHAPTER 9  
HOMEWORK 9-1 . Modern  
Chemistry CHAPTER 9  
HOMEWORK 9-2 (pp. 280282) ..  
write the correct symbols or  
formulas for the word equa nons  
and . 5. aluminum + . modern  
chemistry chapter 8 homework  
8-2 (pp.. p.5: Section 2 Review:  
p.14: Section 3 Review: . Now is  
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...

Modern Chemistry Quiz . Name  
Section Quiz, continued Class  
Date 6. An experiment was  
designed to measure the effect of  
sulfur dioxide ... each sentence or  
best answers each question. 1. A  
chemical reaction was carried out  
three times. The mass of the  
product was 8.93 g for the first  
trial, 8.94 g for the second trial,  
and 8.92 g for the third ...

### **Home - David Brearley High School**

Modern Chemistry 105 Chapter  
Test Name Class Date Chapter  
Test A, continued Use this figure  
to answer questions 7 and 8.

\_\_\_\_\_ 7. A solution containing 35 g

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of  $\text{Li}_2\text{SO}_4$  dissolved in 100 g of water is heated from  $10^\circ\text{C}$  to  $90^\circ\text{C}$ . According to information in the figure, this temperature change would result in a. an additional 5 g of  $\text{Li}_2\text{SO}_4$  in ...

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